

# List of compounds nominated to be detected according to bulk explosive recognition guide including homemade explosives (HME)

As proven at EDA Meeting:

**SOKKS products**  
can be applied as universal training aids



## European Defence Agency (EDA)

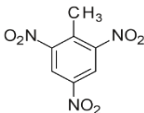
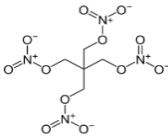
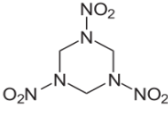
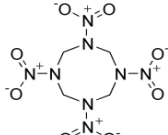
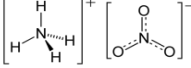
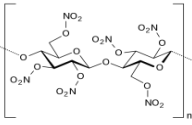
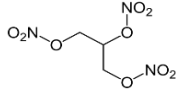
Nato Partner military working dogs counter-IED operations at the Military Center at Kaiser-steinbruch MilHus, in Austria (2015).

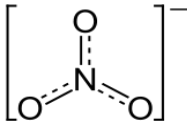
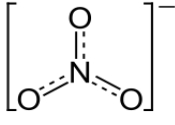
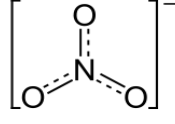
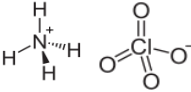
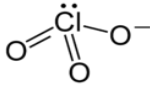
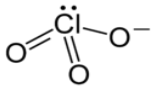
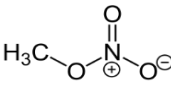
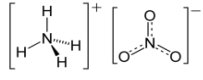
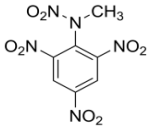
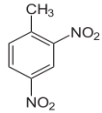
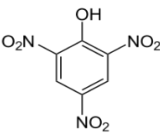
**Important:** Each of the 22 participants from Austrian, Germany, Hungary, Italy, Sweden and the Netherlands **detected SOKKS products** - unknown to the dog handlers - hidden within the test areas!

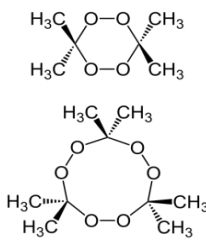
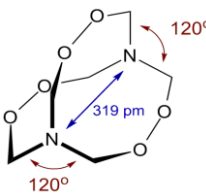


## Compounds detected by SOKKS explosive conditioned canines

(please see comments below the table) by Prof Dr Wolf A Kafka [wolf.kafka@t-online.de](mailto:wolf.kafka@t-online.de)

Name	Synonyms	Formula	Structure	CAS
x TNT	<b>2-Methyl-1,3,5-trinitrobenzene</b> , 2,4,6-Trinitrotoluol, Trinitrotoluen, 2-Methyl-1,3,5-trinitrobenzol, 2-Methyl-1,3,5-trinitrobenzen (IUPAC), 1-Methyl-2,4,6-trinitrobenzol, TNT, Trotyl, AN, Tol, Tolit, Tritol, Tutol	C7H5N3O6		118-96-7
x Pentrite	<b>Nitropenta</b> , Pentaerythryltetranitrat (INN), 1,3-Bis(nitryloxy)-2,2-bis(nitryloxy-methyl)-propan (IUPAC), Pentaerythrittetranitrat, Pentaerythritoltetranitrat, PETN	C5H8N4O12		78-11-5
x Hexogene	<b>Cyclotrimethylenetrinitramin</b> , Cyclonit, T4 und RDX, Perhydro-1,3,5-trinitro-1,3,5-triazin, Hexahydro-1,3,5-trinitro-1,3,5-triazin, Cyclotrimethylenetrinitramin, Cyclonit, RDX, T4	C3H6N6O6		121-82-4
x Octogene	<b>Cyclotetramethylenetetranitramin</b> , Cyclotetramethylenetetranitramin, HMX, LX 14-0, HW 4	C4H8N8O8		2691-41-0
x Nitrat Ammonium	<b>Ammoniumnitrat</b> , Ammonsalpeter, Ammonialsalpeter, brennbarer Salpeter, salpetersaures Ammonium, Ammonnitrat, Ammonium nitricum	H4N2O3		6484-52-2
x Nitrocellulose	<b>Cellulosenitrat</b> , Nitrozellulose, Schießbaumwolle, Blitzwatte	C6H7O11N3		9004-70-0
x Nitroglycerine	<b>Propan-1,2,3-triyltrinitrat (IUPAC)</b> , Trisalpetersäureglycerinester, Glyceriltrinitrat, Trisalpetersäureglycerinester, Trisalpetersäurepropan-1,2,3-triolester, Blasting oil, Glycerinum trinitricum, Trinitroglycerol	C3H5N3O9		55-63-0

x	<b>Poudre Noire</b>	<b>Kaliumnitrat</b> + (selten) <b>Natriumnitrat</b> , Salpeter, Kalisalpeter, E 252,	$\text{KNO}_3(+\text{NaNO}_3)$		
x	<b>Nitrate Potassium</b>		$\text{NaNO}_3$		
x	<b>Nitrate Sodium</b>		$\text{KNO}_3$		
	<b>Perchlorate D'Amonium</b>		$\text{NH}_4\text{ClO}_4$		7790-98-9
x	<b>chlorate de sodium</b>		$\text{NaClO}_3$		7775-09-9
	<b>chlorate de potassium</b>		$\text{KClO}_3$		09.04.3811
	<b>Nitrate de Methylene</b>		$\text{CH}_3\text{NO}_3$		598-58-3
	<b>Nitrate de U'uree</b>	<b>AHL</b> , farblose flüssige Mischung aus <b>Ammoniumnitrat</b> , Harnstoff und Wasser. 7 % Nitratstickstoff, 7 % Ammoniumstickstoff und 14 % Amidstickstoff (aus dem Harnstoff)	$\text{H}_4\text{N}_2\text{O}_3$		
	<b>Tetryl</b>	<b>N-Methyl-N-2,4,6-tetranitroanilin</b> , CE, N-Methyl-N-2,4,6- tetranitroanilin, Trinitrophenylmethylnitramin, Methylpikrylnitramin, Tetralit	$\text{C}_7\text{H}_5\text{N}_5\text{O}_8$		479-45-8
	<b>Dinitrotoluol</b>	2,4-Dinitromethylbenzene, 2,4- Dinitromethylbenzen, 2,4-DNT, Methyldinitrobenzo, Binitrotoluol	$\text{C}_7\text{H}_6\text{N}_2\text{O}_4$		121-14-2
	<b>Melinite</b>	<b>Pikrinsäure, 2,4,6-Trinitrophenol (TNP)</b> , 2,4,6-Trinitrophenol, Trinitrophenol, TNP, Weltersches Bitter	$\text{C}_6\text{H}_3\text{N}_3\text{O}_7$		88-89-1

<b>TATP*</b>	<b>Acetonperoxid (APEX ), IUPAC:</b>			
	<b>3,3,6,6,9,9-Hexamethyl- 1,2,4,5,7,8-hexaoxonan (Trimer), trimeres</b>			
	Acetonperoxid, dimeres	C6H12O4		
	Acetonperoxid, Triacetontriperoxid (TATP), Tricycloacetonperoxid (TCAP), IUPAC: 3,3,6,6-Tetramethyl-1,2,4,5-tetraoxan (Dimer), 3,3,6,6,9,9-Hexamethyl- 1,2,4,5,7,8-hexaoxacyclononan	(Dimer), C9H18O6 (Trimer)		1073-91-2 (Dimer) 17088-37-8 (Trimer)
<b>HMDT*</b>	<b>Hexamethylentriperoxididiamin, 1,6-Diaza-3,4,8,9,12,13-hexaoxa bicyclo[4.4.4]tetradecan (IUPAC)</b>	C6H12N2O6		283-66-9

#### Note:

**Compounds marked by x belong to relevant military explosives, as listed within the internal SOKKS instructions. However, as reported by different users, SOKKS conditioned canines detect at least all of the listed by Munitique DEVIS**

Whilst contamination/purity is open in compounds offered by Munitique DEVIS SOKKS products are based on high level purity compounds.

Handling of compounds in g units (**merely spoon full amounts as offered - at very high price levels - by Munitique DEVIS**) are of less practical value: Contamination of hiding places, loss of material, etc.

**For examinations under "real" field-conditions it seems therefore advantageous** to apply explosive-asservates commonly applied in industrie and military operations.

#### In addition:

**Examples of further primary high explosives - most of them not in the list offered by Munitique DEVIS - are detected by SOKKS conditioned canines (personal army reports).**

## Note in addition:

SOKKS explosive conditioning material is also directed for the detection of TATP (= APEX) (trimeric Acetonperoxide) by SOKKS explosive conditioned dogs.

This has been confirmed by military and police sectors (including the GIGN). Since handling of "sharpened" (= dried from water) TATP is extremely impact-, touch and heat sensitive, it is still open, however, of whether the dogs detected TATP or its precursors.

On base of this, TATP production will commonly be "started" shortly before application by mixing up the appropriate starting materials (for example: [acetone + hydrochloric acid] + hydrogen peroxide).

This may be seen to be confirmed by the fact that the terrorists in Brussels insisted on carrying their bags by themselves into the taxi. (It seems to be proven that terrorists used TATP in Brussels and also in Paris).

Nevertheless SOKKS explosive conditioned dogs should detect as well "sharp" TATP as TATP in precursor state.

Recently developed technical TATP sensors (based on differential micro-weight measurements of TATP-molecules annealed to specially pretreated surfaces), however, up to now, they do not reach the sensitivity of the dog's nose.

Chemicals::

- Acetone 50 % ( conc. )• Hydrogenperoxide 50 % ( 30 % )
- hydrochloric acid 30 % from total volume ( 32 % )
- Cooling bath (for the conservation of 0 oC reaction temperature)
- Water for washing

Implementation:

- acetone and hydrochloric acid unite with stirring and at 0 degrees Celsius
- Slowly add hydrogen peroxide
- Acetone precipitates white

Special instructions:

- Fabric sublimated ; difficult to ensure safe storage ; drives plug from reservoir (bottle)
- must be dry to explode ; soaked by the fuse
- by sublimation is increased , the water concentration
- remaining hydrochloric acid within the material leads to the development of heat during wrapping in aluminum - foil

# Explosives most relevant for the German Bundespolizei

## Explosives

Trade name	Chemical nomenclature	Molecules *10 <sup>10</sup> /cm <sup>3</sup>	formula
Ammongelit	Ammoniumnitrat + Nitroglycol	70000000	NH <sub>4</sub> NO <sub>3</sub> + C <sub>5</sub> H <sub>8</sub> N <sub>2</sub> O <sub>6</sub>
Ammonsalpeter	Ammoniumnitrat	70000000	NH <sub>4</sub> NO <sub>3</sub>
<b>Hexogen (RDX)</b>	Cyclotrimethylen-trinitramin	40	C <sub>3</sub> H <sub>5</sub> N <sub>6</sub> O <sub>6</sub>
Kaliumchlorat	Kaliumchlorat	100000	KClO <sub>3</sub>
Kalisalpeter	Kaliumnitrat	10000	KNO <sub>3</sub>
Natriumchlorat	Sodium Chlorat	100000	NaClO <sub>3</sub>
Nitroglycerin	Trinitroglycerin	70000000	C <sub>3</sub> H <sub>5</sub> N <sub>3</sub> O <sub>9</sub>
Nitropenta (PETN)	Pentaerythritol tetranitrat	100	C <sub>2</sub> H <sub>5</sub> N <sub>4</sub> O <sub>12</sub>
Nitrozellulose (NC)		30	
<b>Oktogen</b>	Cyclotetramethylen-tetranitramin <b>HMX</b>	<b>1</b>	<b>C<sub>4</sub>H<sub>8</sub>N<sub>8</sub>O<sub>8</sub></b>
Blackpowder	Kaliumnitrat + Schwefel + Holzkohle	10000	KNO <sub>3</sub> + S + C
Treibladungspulver	Nitrozellulose + -glycerin + -guanidin	400000	
Trinitrotoluol (TNT)	<b>2,4,6-Trinitrotoluol</b>	<b>60000</b>	<b>CH<sub>2</sub>H<sub>3</sub>N<sub>3</sub>O<sub>6</sub></b>

**Explosives: Molecules\*10<sup>10</sup>/cm<sup>3</sup> ppm-values (Meyer, 1985 & EMC, Consulting Services Dr Hoffmann, Schramberg) due to calculations via Loschmidt-Konstant (N=2,686\*10<sup>25</sup>/m<sup>3</sup>). e.g. 1 Molecule Octogen in 10<sup>9</sup> Molecules air. SOKKS-explosive Material is nearly ca. x10000 less down !**

**A detecting dog would thus detect a dilution factor of 1 to 10<sup>13</sup>, he would detect 1 ml octogen in an air volume of a big lake 20x20x0,07 km.**

Additional examples of primary high explosives are:

Acetone peroxide  
Ammonium permanganate  
Azo-clathrates  
Chlorine azide  
Copper acetylide  
Cyanogen azide  
Diazodinitrophenol  
Dichlorine heptoxide  
Disulfur dinitride  
Hexamethylene triperoxide diamine Lead azide  
Hypofluorous acid  
Lead styphnate  
Lead picrate<sup>[4]</sup>  
Mercury(II) fulminate  
Nitrogen trichloride  
Nitrogen triiodide  
Nitroglycerin  
Octaazacubane  
Silver azide  
Silver acetylide  
Silver fulminate  
Sodium azide  
Tetraamine copper complexes Tetraazidomethane  
Tetrazine  
Tetranitratocarbon  
Tetrazoles  
Xenon oxytetrafluoride  
Xenon tetroxide  
Xenon trioxide